



**National
Metrology
Institute**

VSL



**Results of interlaboratory
comparison with gaseous
reference materials**

Webinar Metrology for indoor air quality

Iris de Krom, VSL

11 April 2024 – Online

Introduction

- VOC gaseous reference materials
 - Production PRM & CRM
 - Analysis
 - Uncertainty
 - Storage and stability
 - Conclusion
- Interlaboratory comparison
 - Comparison
 - Results
 - Conclusion



Relevant VOCs

- VOCs from according to EN 16516 (8.2.2.3)
 - *n*-Hexane
 - MIBK (methyl isobutyl keton)
 - Toluene
 - Butyl acetate
 - Cyclohexanone
 - *o*-Xylene
 - Phenol
 - 1,3,5-trimethylbenzene
 - ~~n~~-Hexadecane (not suitable for preparation of static gas mixtures)
- 1,2,3-trimethylbenzene: pure component has low purity (80%)
 - Impurities: 1,3,5-trimethylbenzene, 1,2,4-trimethylbenzene
 - GC-analysis: possible overlap with phenol & bias due to impurities

Preparation Primary Reference Materials (PRM)

Static gas mixtures (ISO 6142)

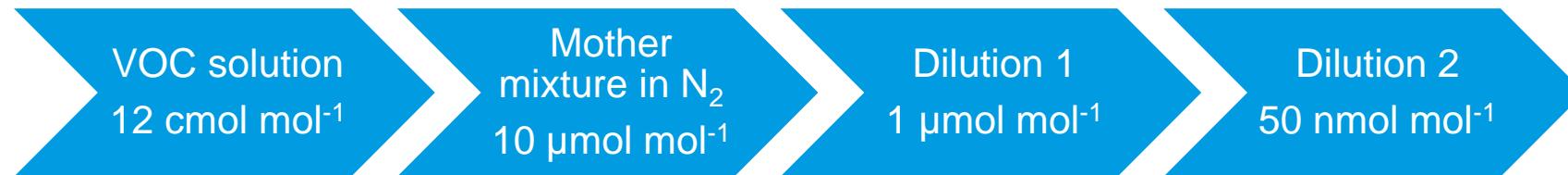


Dynamic gas mixtures (ISO 6145)



Preparation Primary Reference Materials (PRM)

- Static gas mixture preparation (ISO 6142)



- Dynamic gas mixture preparation
 - Continuous injection (ISO 6145-4)



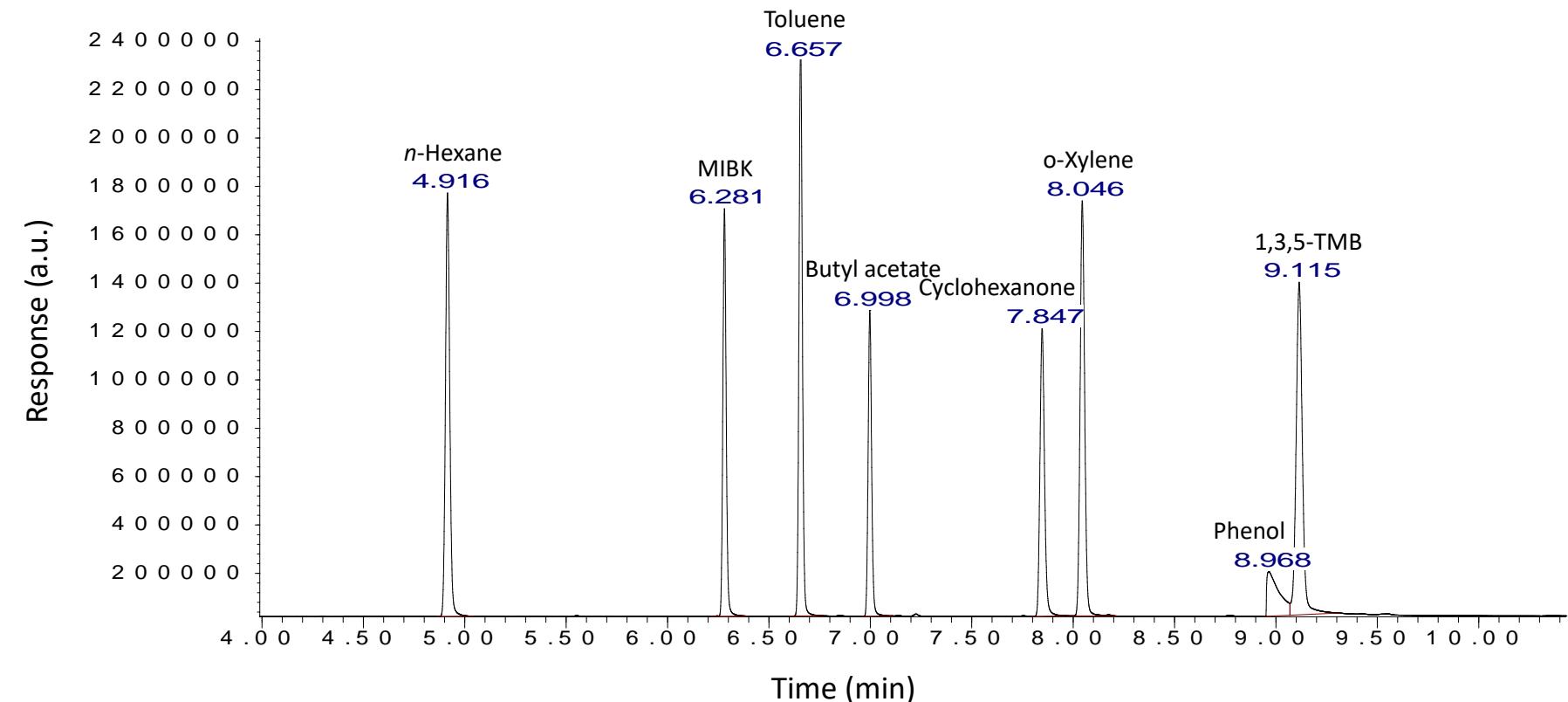
Preparation Calibrated Reference Materials (CRM)

- Dynamic gas mixture preparation
- Sorbent tube sampling → pumped sampling (ISO 16017-1)
 - Tenax TA (TTA)
 - Materials Emissions/Soil Gas Monitoring (ME)
- Nominal mass sampled into tubes was 125 ng per VOC
- Develop and validate analytical method
- Compare static and dynamic gas mixture preparation
- Stability study of CRM



Analysis sorbent tubes

- TD-GC-FID
- DB-1, 30 m long, 0.32 mm internal diameter, 1.00 μm film thickness



Validation tube sampling & analysis method

- Breakthrough test sorbent tubes
 - TTA → *n*-hexane → 1.9%
 - ME → no breakthrough was observed
- Desorption efficiency test sorbent tubes
 - Full desorption for all components and sorbents
- Precision: repeatability (s_r) and reproducibility (s_R) → ANOVA calculation
 - TTA sorbent tubes

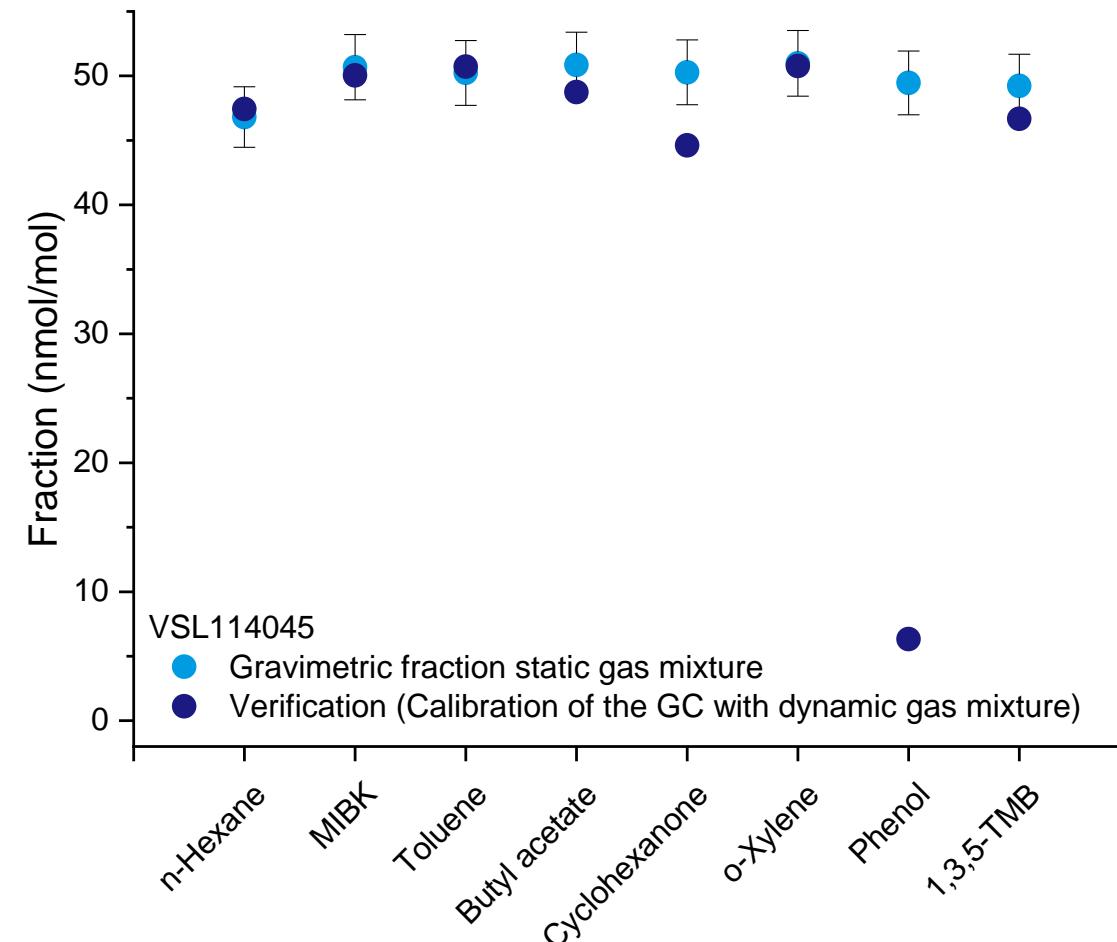
	<i>n</i> -Hexane	MIBK	Toluene	Butyl acetate	Cyclohexanone	<i>o</i> -Xylene	Phenol	1,3,5-TMB
s_r	1.66%	0.97%	0.62%	2.03%	3.43%	0.64%	11.03%	1.17%
s_R	2.57%	1.34%	1.16%	2.22%	3.50%	1.26%	14.69%	1.92%

- ME sorbent tubes

	<i>n</i> -Hexane	MIBK	Toluene	Butyl acetate	Cyclohexanone	<i>o</i> -Xylene	Phenol	1,3,5-TMB
s_r	0.64%	0.74%	0.70%	1.51%	2.51%	0.61%	6.19%	0.51%
s_R	1.17%	1.11%	1.24%	1.60%	2.51%	1.12%	9.89%	1.47%

- Measurement uncertainty

Comparison static and dynamic gas mixtures



Uncertainty static PRM

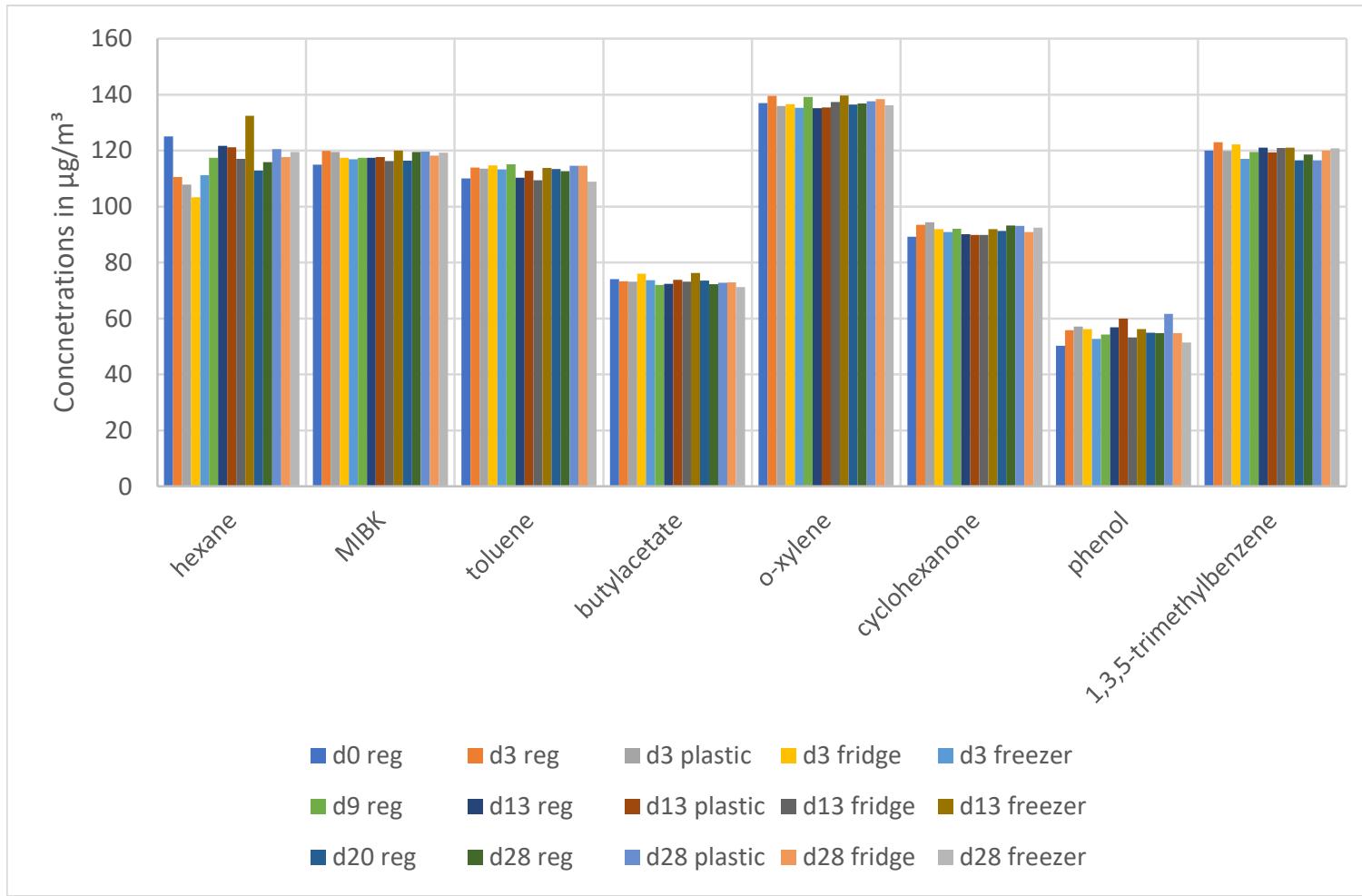
- Uncertainty sources
 - Preparation and sampling into sorbent tubes → 1% ($k = 2$)
 - Calibration GC with the dynamic PRM → 5% ($k = 2$)
 - Measurement uncertainty ($k = 2$)

VOC	U_{meas} (%)	U (%)
<i>n</i>-hexane	2.4	6
MIBK	2.2	6
Toluene	2.4	6
Butyl acetate	3.2	6
Cyclohexanone	5	7
<i>o</i>-Xylene	2.2	6
Phenol	20	20
1,3,5-TMB	3.0	6

Storage and stability

- CRM → TTA sorbent tubes
 - Storage
 - Room temperature
 - Fridge
 - Freezer
 - Packaging
 - Regular
 - Dark packing
 - $t = 0, 3, 9, 13, 20 and 28 days$

Stability study CRM



Conclusion preparation PRM and CRM

- PRM and CRM developed with selected VOCs from EN16516
 - Phenol decomposes or precipitates in a cylinder.
 - Dynamic gas mixture with phenol can be obtained
- Successful sampling into sorbent tubes
 - Keep in mind the save sampling volume of *n*-hexane when using TTA
 - Used for calibration of GC or other measurement systems
 - Used for interlaboratory comparison
- Uncertainty
 - Dynamic PRM 5 %
 - Static PRM 6 % - 20 %
- Stability CRM tested 1 year
 - 28 days at room temperature in ME and TTA sorbent tubes

Report

- Published
 - D3 [Guideline for the preparation and analysis of gPRMs and gCRMs of indoor air pollutants stated in the EU-LCI list with relative uncertainties below 5 % \(k = 2\) and a shelf life of at least 1 year \(zenodo.org\)](#)

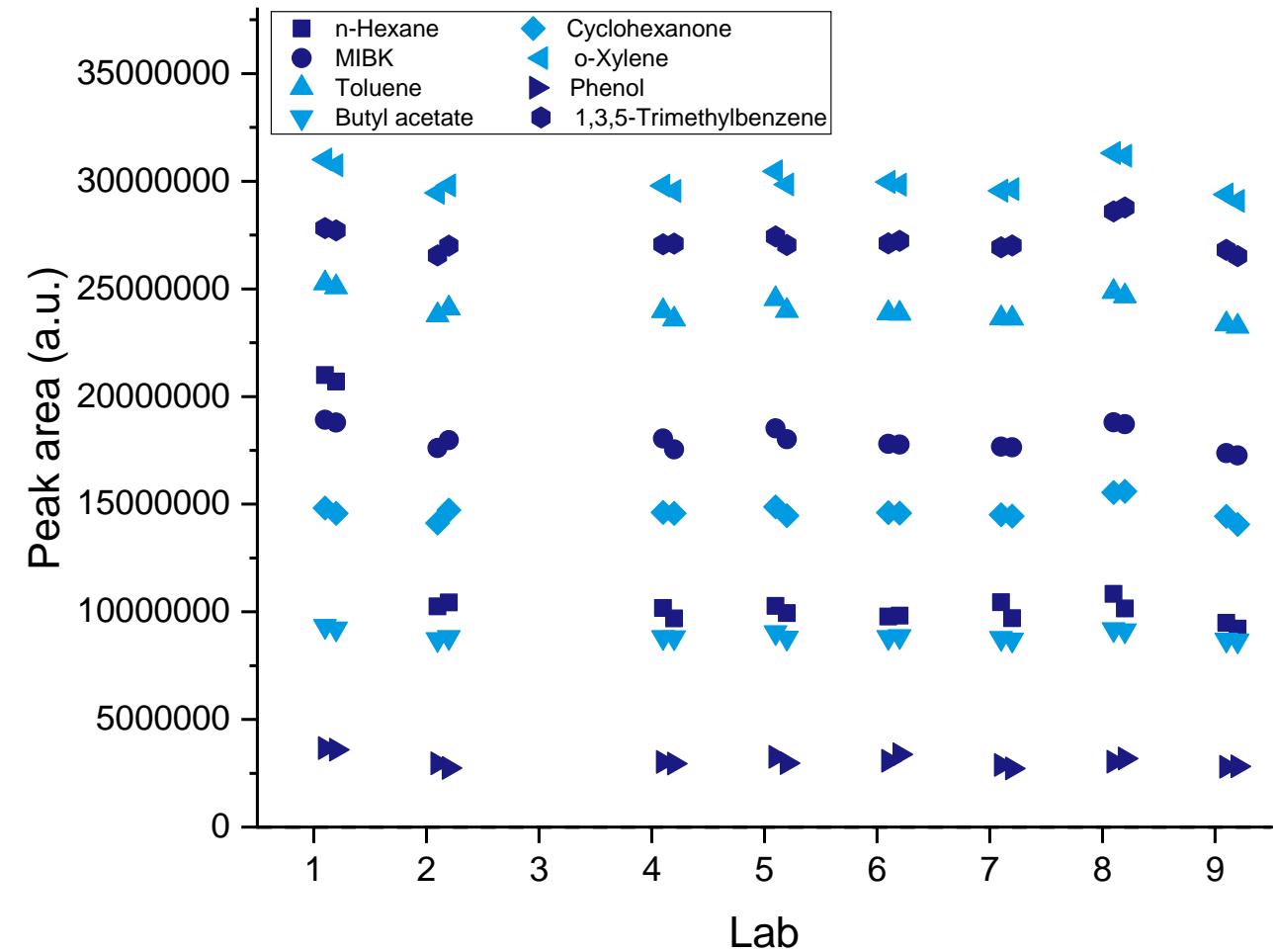
Interlaboratory comparison

- Goal
 - Assess the participants capabilities in sampling and measuring VOCs
 - *n*-Hexane, MIBK, Toluene, Butyl acetate, Cyclohexanone, o-Xylene, Phenol, 1,3,5-Trimethylbenzene
 - External validation gCRM
- Workshop
 - Participants sample at least 3 tubes with dynamic gas mixture of VOCs in air at VITO
 - VSL determined the reference value
 - 4 participants
- Interlaboratory comparison
 - VITO sampled 3 participants tubes with dynamic gas mixture
 - VSL determined the reference value
 - 8 participants



Homogeneity samples interlaboratory comparison

- Lab 1
 - Tubes were resampled
 - Different reference value
- Lab 3
 - Only participated in the workshop
- Results other labs
 - comparable within the uncertainty of 5 %
- n-Hexane
 - Results too low



Data processing

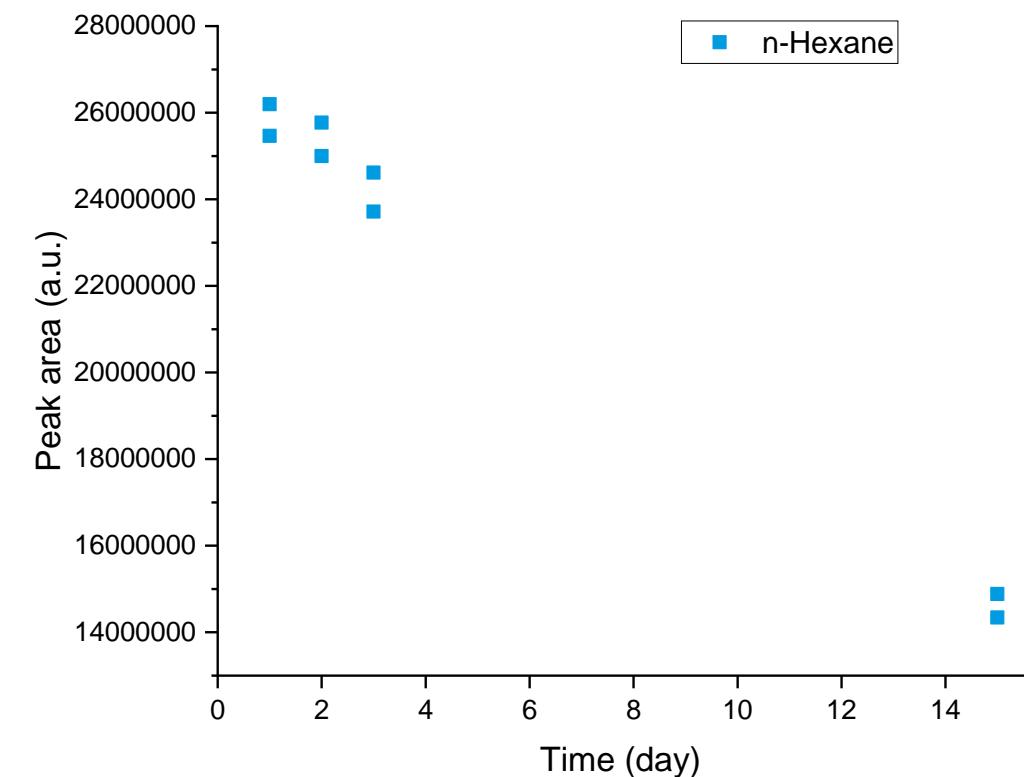
- Zeta scores for the assessment of laboratory results according to ISO 13528

$$\zeta_{lab(i)} = \frac{m_{lab(i)} - m_{ref(i)}}{\sqrt{u_{lab(i)}^2 + u_{ref(i)}^2}}$$

- $m_{ref(i)}$ → reference value VSL

- *n*-Hexane

- Interlaboratory comparison: Consensus value
 - Workshop: Reference value



Consensus values

Workshop

Component	m	u(m)	m _{ref}	U(m _{ref})	T
n-Hexane	122	21	107	5	28
MIBK	134	15	138	7	22
Toluene	124	14	128	6	20
Butyl acetate	84	9	86	4	12
Cyclohexanone	114	8	122	6	0.0
o-Xylene	154	10	162	8	7
Phenol	69	4	99	5	0.0
1,3,5-TMB	131	11	138	7	13

Interlaboratory comparison

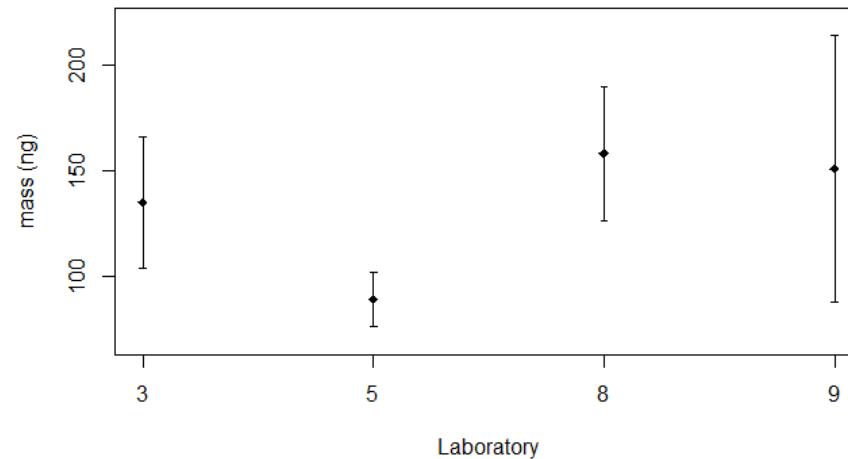
Component	m	u(m)	M _{ref} ¹	U(m _{ref})	T
n-Hexane	98	6	49.5 ²	2.5 ²	8
MIBK	99	7	119	6	15
Toluene	97	7	110	5	14
Butyl acetate	101	7	74	4	14
Cyclohexanone	99	8	106	5	17
o-Xylene	107	4	139	7	4
Phenol	26	7	61	3.0	13
1,3,5-TMB	94	6	119	6	9

¹ Save for Lab 1

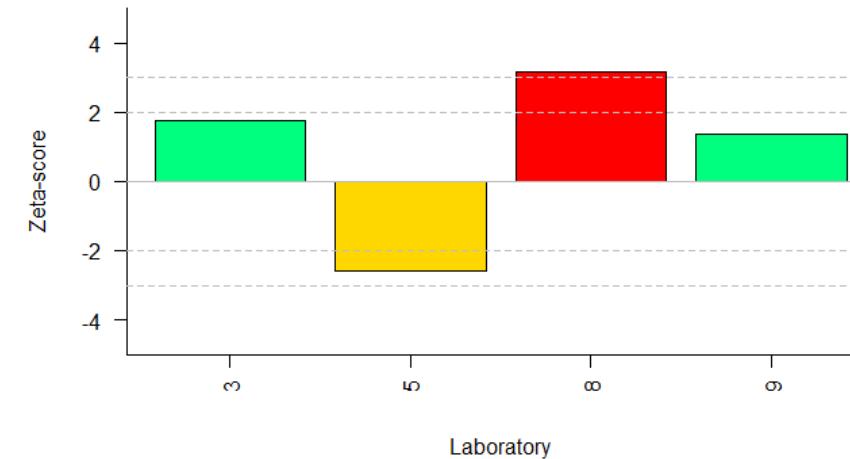
² Consensus value used

Preliminary results workshop

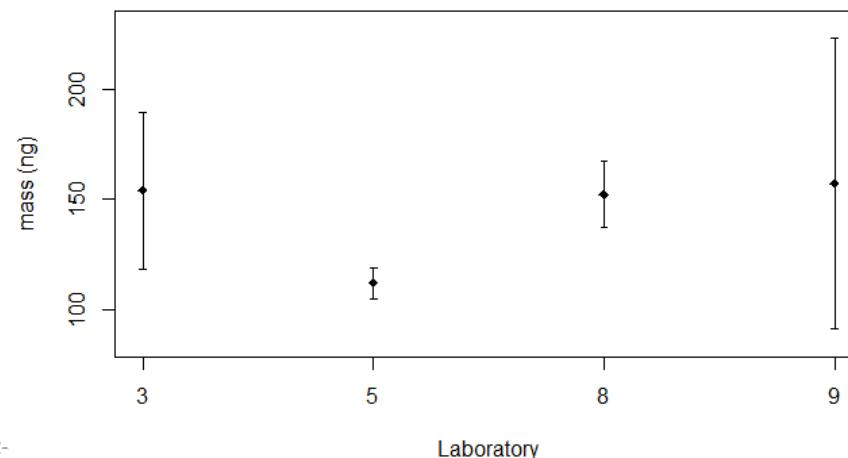
Reported data for n-hexane



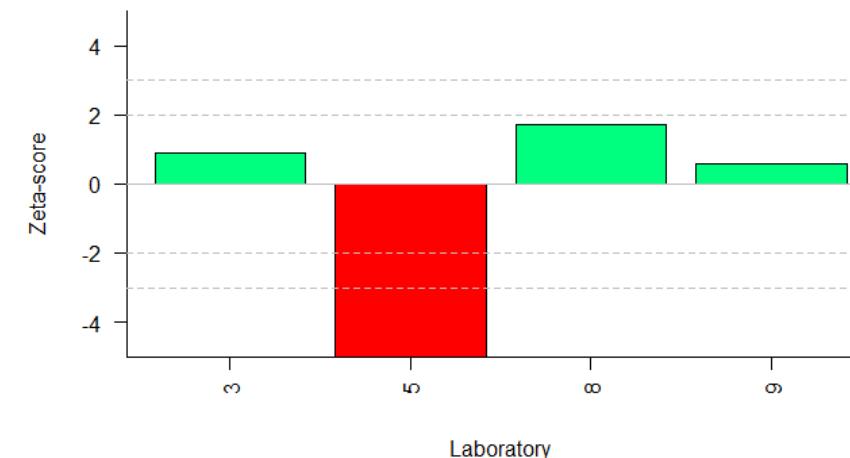
Zeta scores for n-Hexane



Reported data for methyl isobutyl ketone

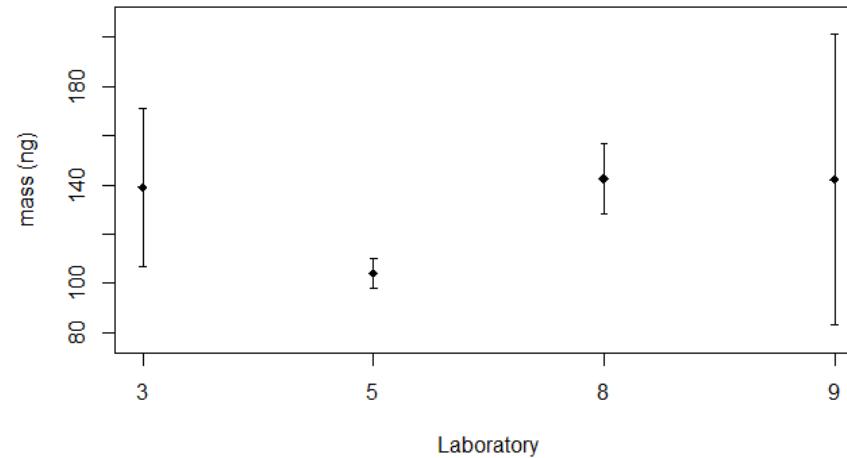


Zeta scores for methyl isobutyl ketone

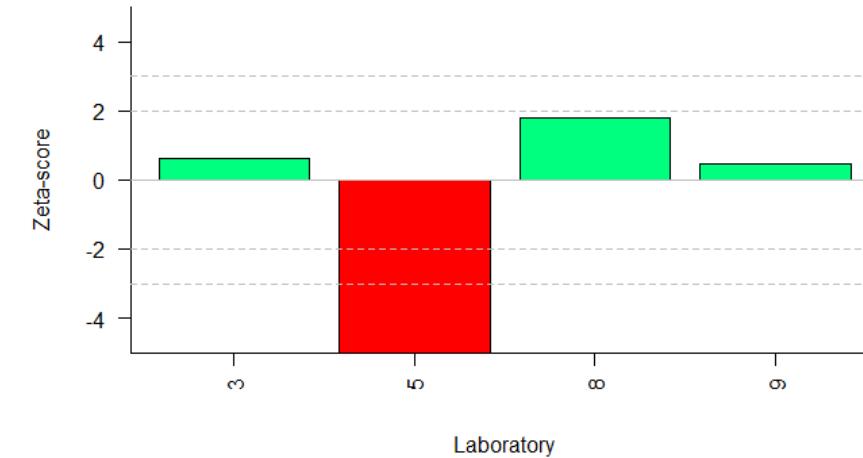


Preliminary results workshop

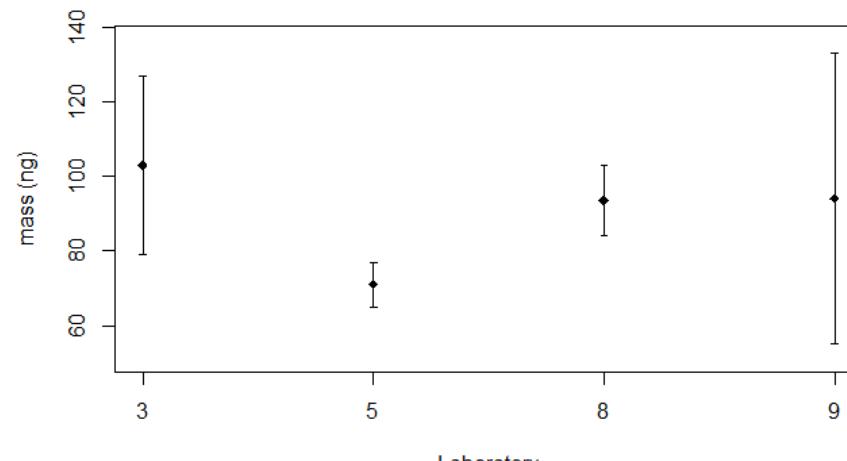
Reported data for toluene



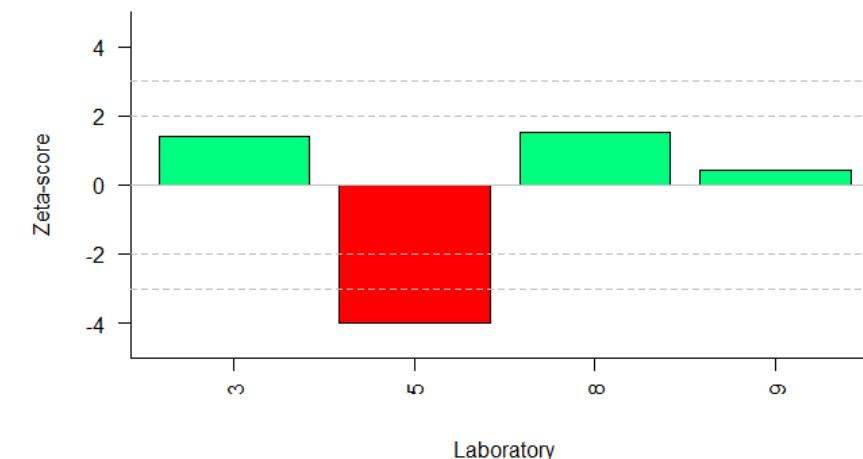
Zeta scores for toluene



Reported data for butyl acetate

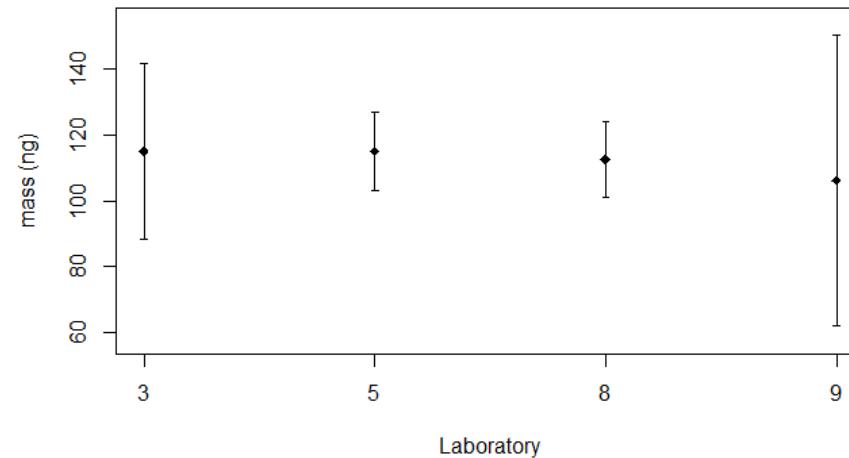


Zeta scores for butyl acetate

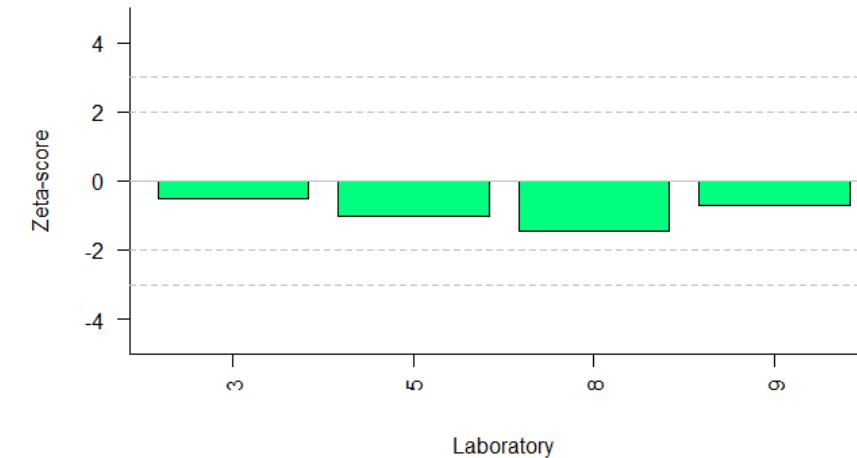


Preliminary results workshop

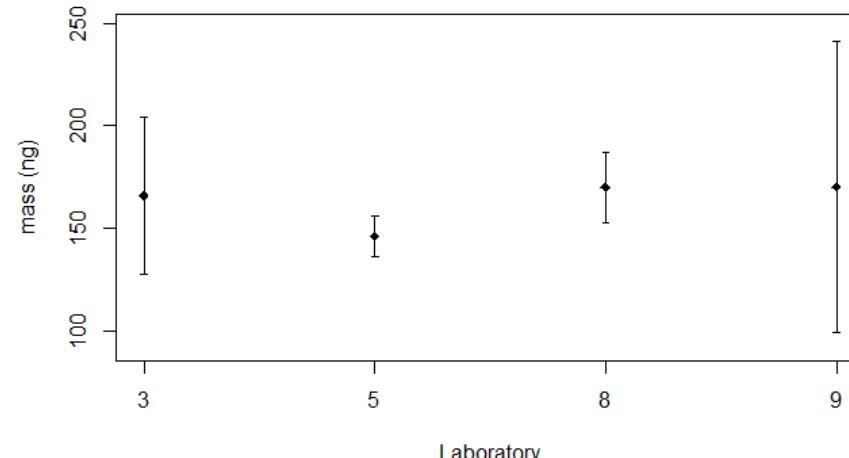
Reported data for cyclohexanone



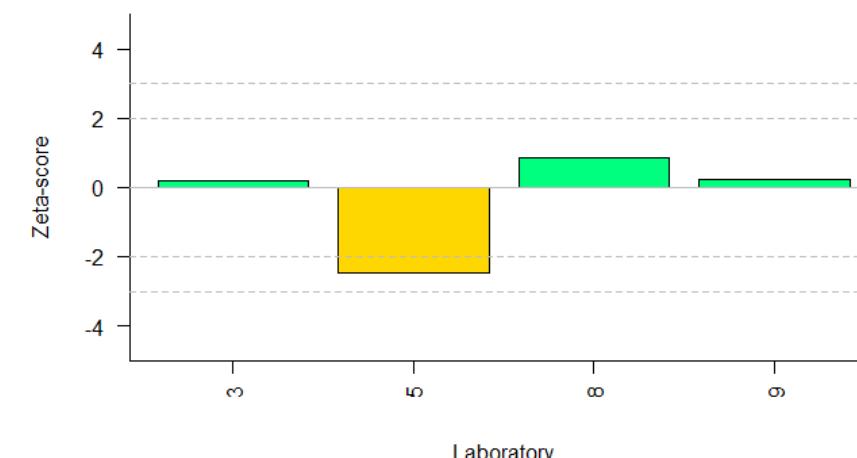
Zeta scores for cyclohexanone



Reported data for o-xylene

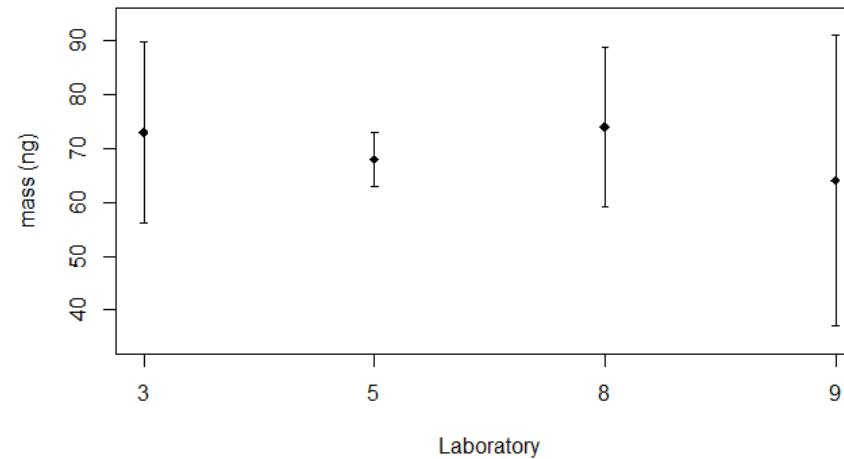


Zeta scores for o-xylene

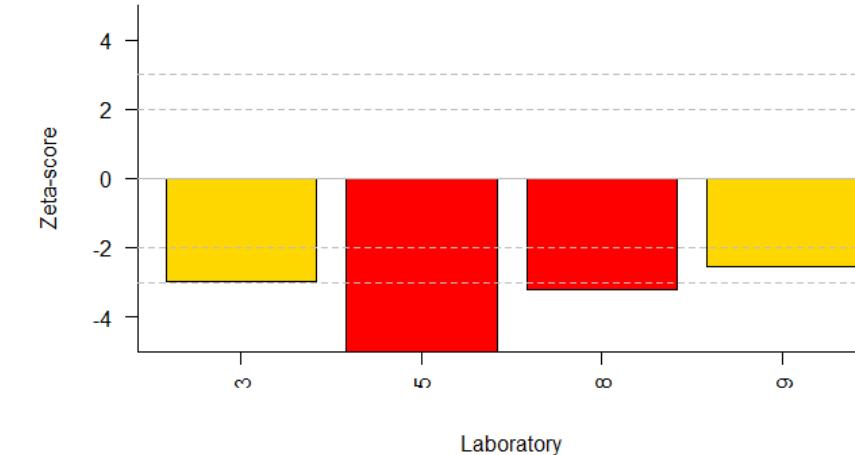


Preliminary results workshop

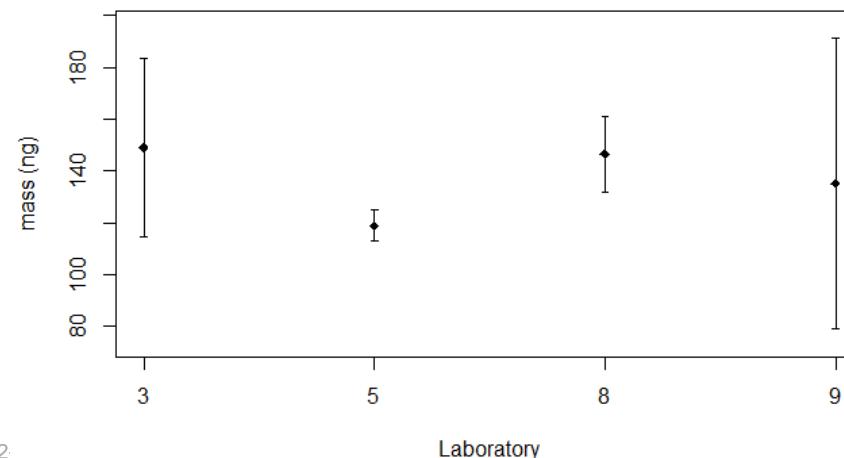
Reported data for phenol



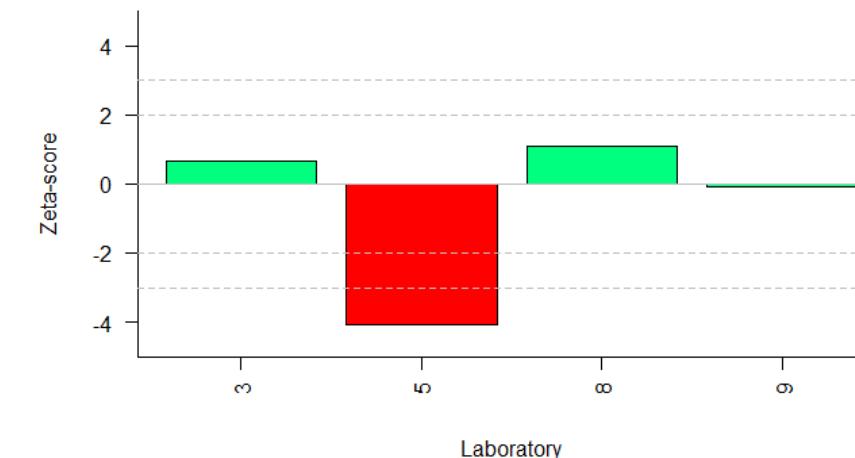
Zeta scores for phenol



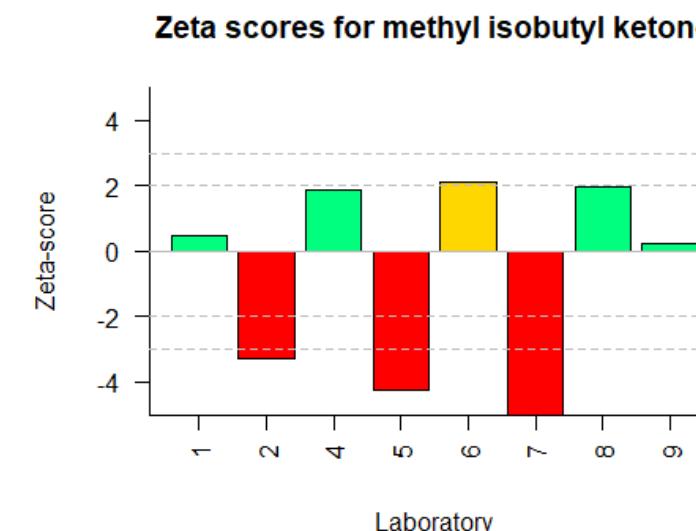
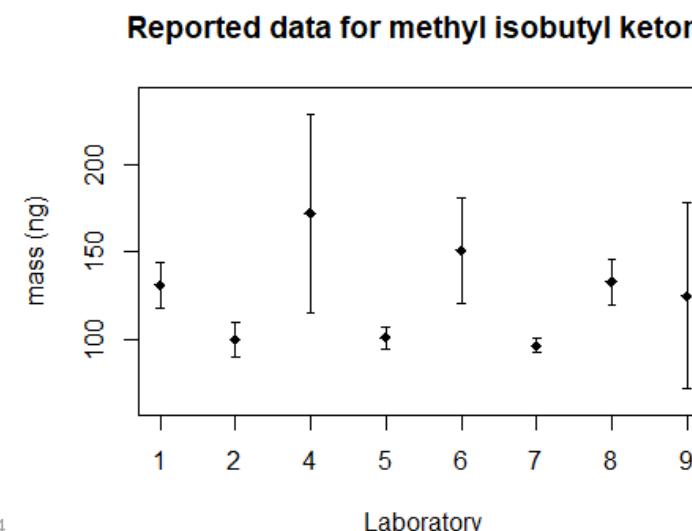
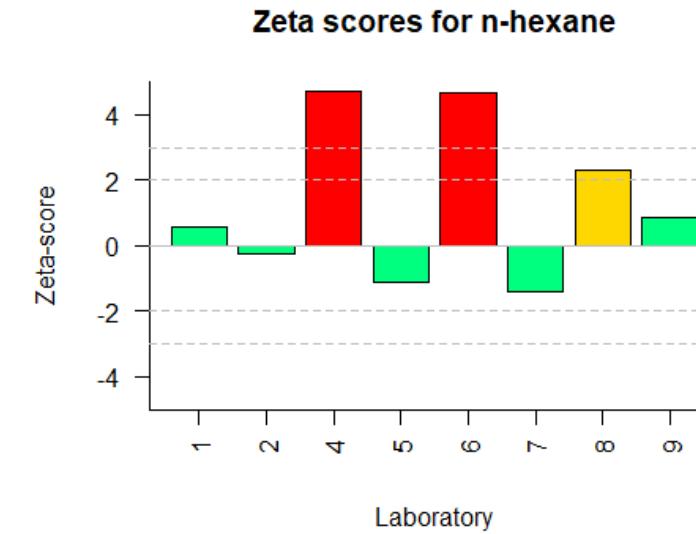
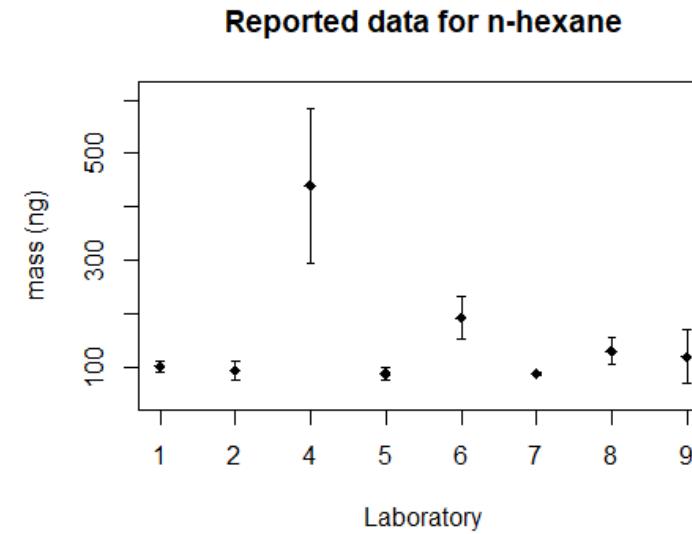
Reported data for 1,3,5-trimethylbenzene



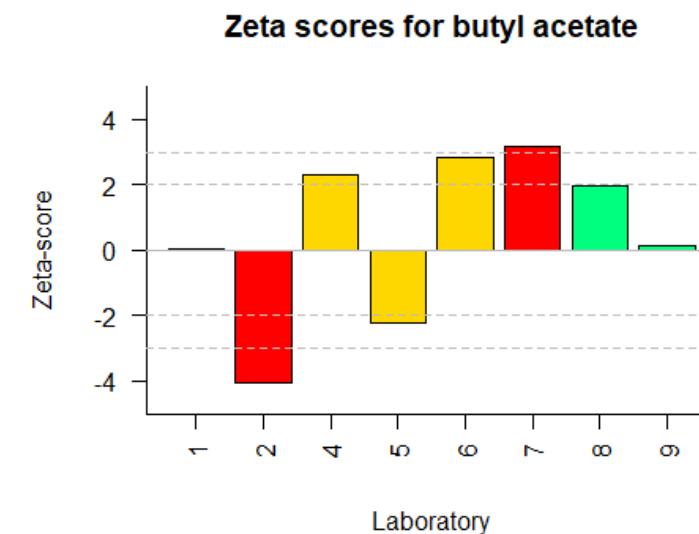
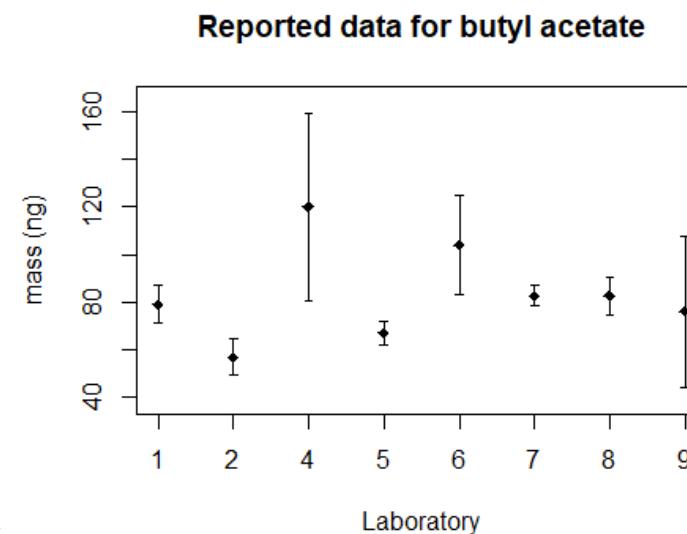
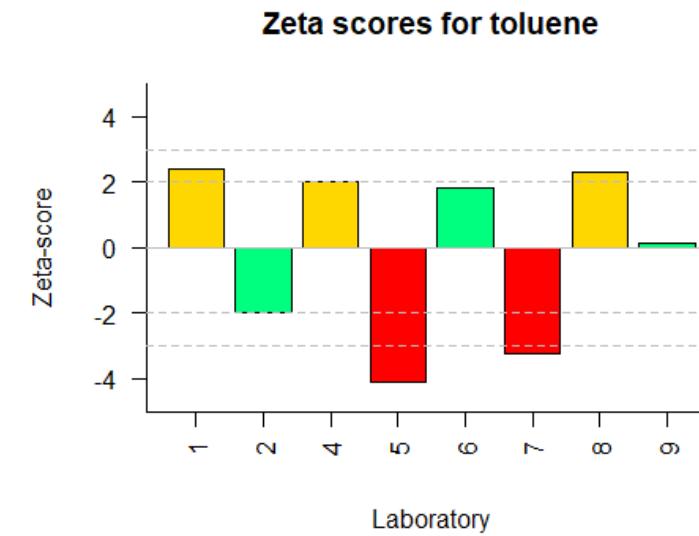
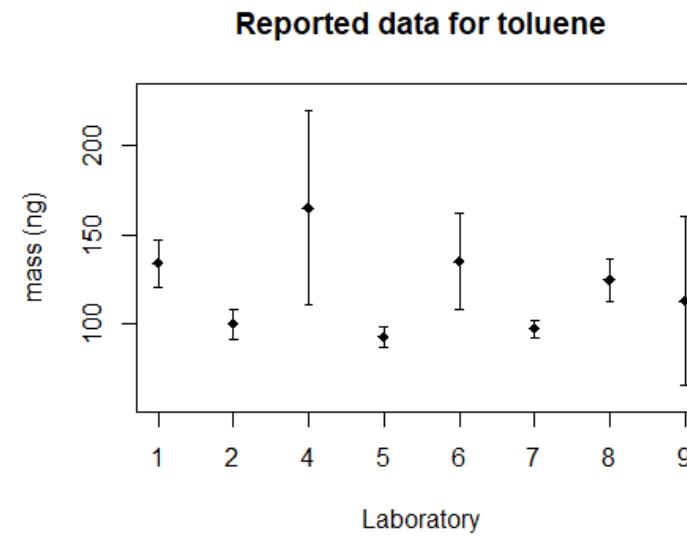
Zeta scores for 1,3,5-trimethylbenzene



Preliminary results interlaboratory comparison

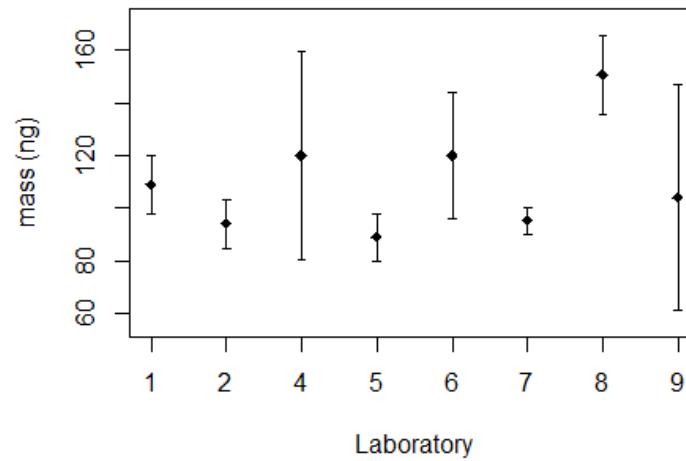


Preliminary results interlaboratory comparison

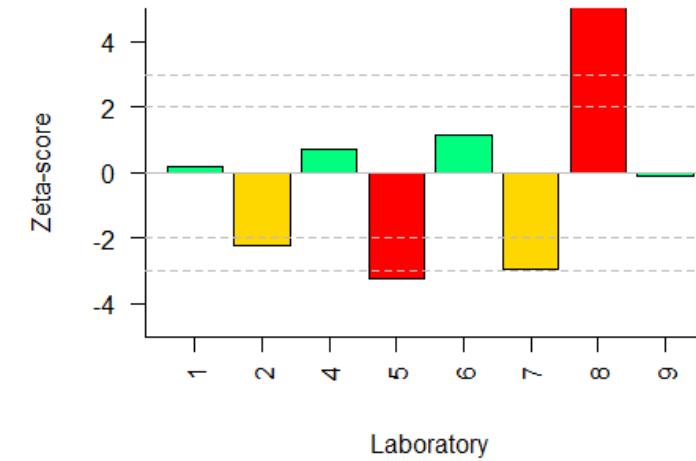


Preliminary results interlaboratory comparison

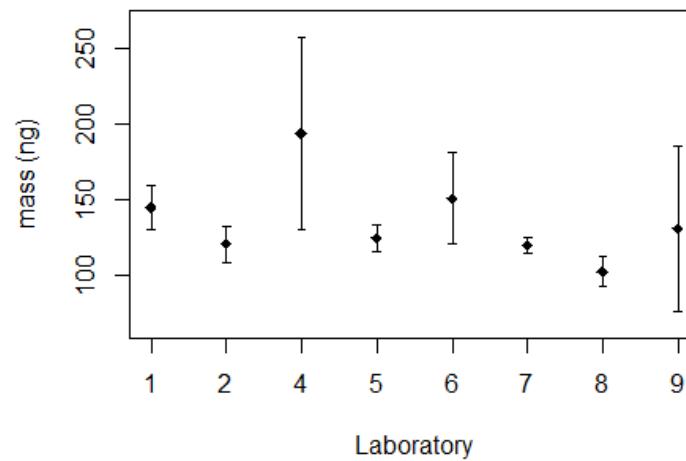
Reported data for cyclohexanone



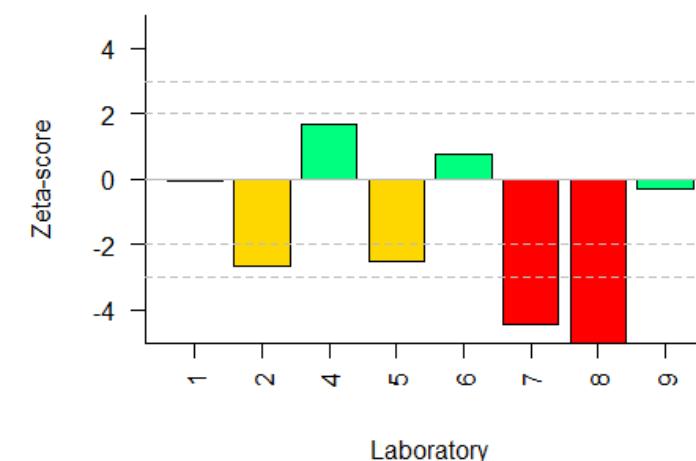
Zeta scores for cyclohexanone



Reported data for o-xylene

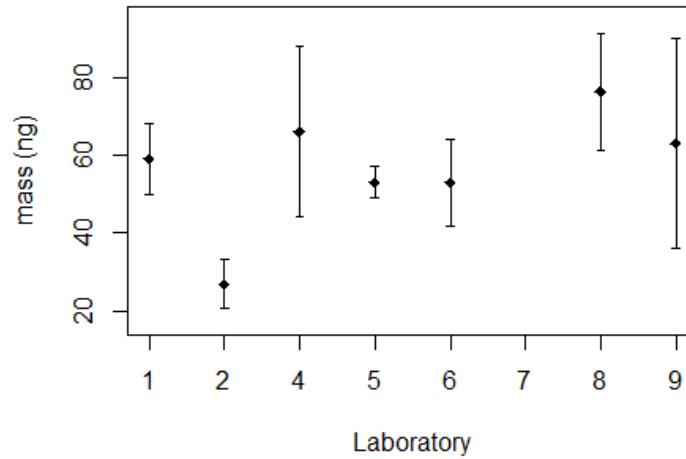


Zeta scores for o-xylene

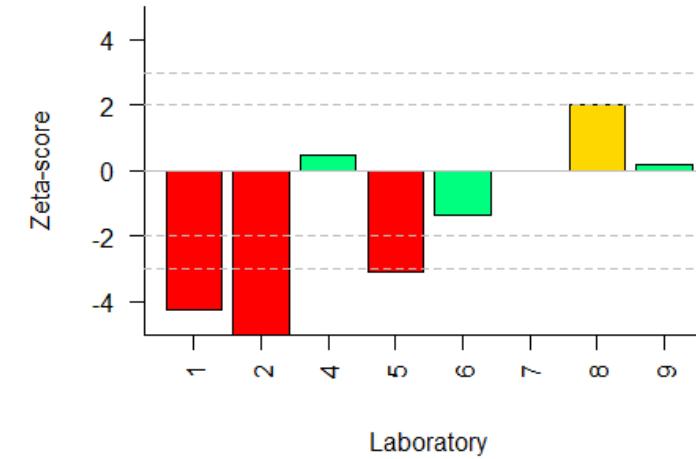


Preliminary results interlaboratory comparison

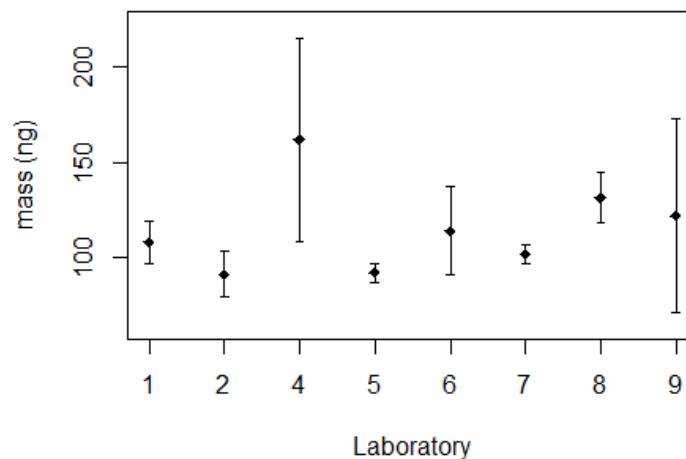
Reported data for phenol



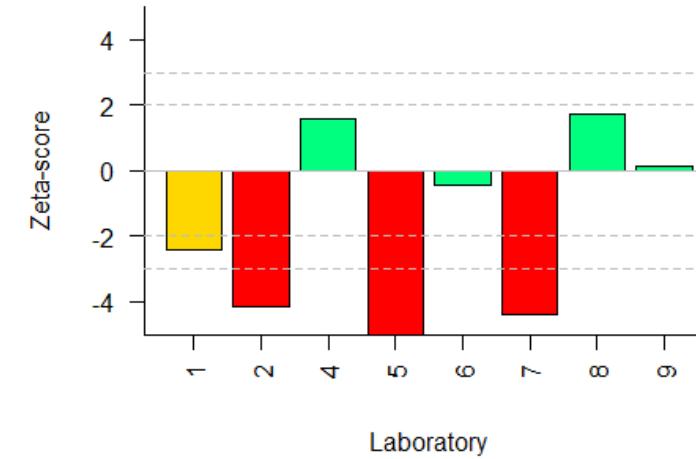
Zeta scores for phenol



Reported data for 1,3,5-trimethylbenzene



Zeta scores for 1,3,5-trimethylbenzene



Conclusion

- The results highlight the complexity of these measurements
- Workshop:
 - Satisfactory results for labs 3, 8 and 9
 - Lab 5 reported biased results for most VOC
- Interlaboratory comparison
 - Lab 9 performed satisfactory for all VOCs
 - Labs 1, 4 and 6 perform satisfactory for most VOCs and questionable for a few components
 - Lab 2, 5, 7 and 8 report biased results for most VOCs

Reports

- Available in the coming months
 - *D5: "Summary report on the evaluation of the inter-laboratory comparison performed with the ERM and gCRMs: including information on the repeatability and reproducibility of the data and an evaluation of the major uncertainty sources in sampling, analysis and reporting, according to the EN 16516 test method".*
 - *D6: "Report on the preparation and measurement uncertainty of ERMs, gPRM and gCRMs"*

Contact Info:
Iris de Krom
idekrom@VSL.nl

